

0040-4020(95)00359-2

Base-Induced Conversion of 5-Aminothiazolium Salts into Substituted Pyrroles and Pyrrolines via 1,3-Dipolar Cycloadditions with Electron-Deficient Alkynes and Alkenes.

Fabienne Berrée, and Georges Morel *

Laboratoire de Physicochimie Structurale, URA CNRS nº 704, Campus de Beaulieu, 35042 Rennes, France.

Abstract: A series of mesoionic thiazoles 2 were generated and converted in situ by treating the 5-aminothiazolium chlorides 1 with several electrophilic dipolarophiles in the presence of DBN or NEt3. The reactions involved an 1,3-dipolar cycloaddition of the "masked" cyclic azomethine ylides across the olefinic or acetylenic π -bond, yielding unstable N-bridged adducts as the first step. Thus, DMAD and methyl propiolate gave functionalized pyrroles via a subsequent extrusion of isothiocyanate. Dimethyl fumarate, maleate and fumaronitrile afforded a variety of polysubstituted pyrrolines, pyrroles and aminobutadienes through base-catalyzed rearrangements of the primary 1:1 cycloadducts, with retention of the elements of isothiocyanate. A similar initial cycloadduct 17 was isolated from N-phenylmaleimide which generally led to the condensed imino thiopyrans 20. Some of the 1,3-dipolar reactional stereoselectivities were deduced from the stereochemistry or distribution of the final products and found to be markedly dependent on the substituent groups present.

Introduction. A few synthetic routes to 5-aminothiazolium salts and corresponding mesoionic thiazolium-5-aminides have been described about twenty years ago but their chemistry has not been extensively explored ¹. In a recent article, we showed that such compounds are now easily accessible via three-component condensations using isocyanides as cyclization reagents.² In particular, we described the one-pot and efficient preparation of the 5-tert-butylamino-4-phenylthiazolium chlorides 1a, b from a mixture of aryl chlorothioformate, N-benzylidenemethylamine and tert-butyl isocyanide. Similarly, the treatment of imino chlorosulfides with dimethylthioformamide and isocyanides at room temperature provides selectively the 4,5-bis-aminothiazolium salts 1c-1. ²

 \mathbf{R}^{t}

				R ² -N	, S ⊕ R ⁴	CI NH	⊖ R³			
	R^1	\mathbb{R}^2	R ³	R^4			\mathbb{R}^1	R ²	\mathbb{R}^3	\mathbb{R}^4
a	PhS	Me	t-Bu	Ph	g		MeS	CH2CO2Et	t-Bu	Me ₂ N
b	p-Tol O	Me	t-Bu	Ph	h		PhS	CH2CO2Et	t-Bu	Me ₂ N
С	MeS	CH ₂ Ph	t-Bu	Me ₂ N	i		MeS	CH ₂ PO(OEt) ₂	CH ₂ Ph	Me ₂ N
d	MeS	CH ₂ Ph	CH ₂ Ph	Me ₂ N	j		MeS	i-Pr	t-Bu	Me ₂ N
е	MeS	CH ₂ Ph	CH2CO2Et	Me ₂ N	k		MeS	i-Pr	CH ₂ Ph	Me ₂ N
f	PhS	CH ₂ Ph	t-Bu	Me ₂ N	ı		MeS	2,6-Me ₂ C ₆ H ₃	t-Bu	Me ₂ N

1,3-Dipolar cycloaddition reactions involving a wide variety of mesoionic ring systems and dipolarophiles have been reported in the literature and used for the synthesis of a large number of five-membered heterocycles.³ It has been demonstrated that these methods proceed through unstable 1:1 primary cycloadducts which decompose to more stable products with the evolution of small molecules such as carbon dioxide or disulfide, carbonyl sulfide, isocyanates or isocyanic acid and also carbodiimides. For instance, the reaction of münchnones and azlactones with acetylenic dipolarophiles was shown to constitute a pyrrole synthesis of very broad scope ⁴ through the loss of CO₂. In some cases (i.e. intramolecular münchnone ⁵, isomünchnone and thioisomünchnone ⁶ cycloadditions) isolated compounds corresponded to the initial cycloadducts, without retro-Diels-Alder process. All these studies have resulted in practical, unique syntheses of numerous functionalized monocyclic and ring annulated heterocycles. However, in comparison with the widespread use of thiazolium-5-thiolates and thiazolium-4 or 5-olates, ^{3,6,7} similar applications of mesoionic thiazolium-5-aminides are lacking.⁸ Moreover, a study of such reactions involving the loss of isothiocyanate has found very little attention.⁹

Salts 1 are potential cyclic azomethine ylides and we decided to examine their dipolar reactivity. We report here their chemical behaviour in a basic medium containing a standard acetylenic or olefinic dipolarophile. Mesoionic thiazoles 2 are not stabilized by an electron-withdrawing substituent group R³. They can never be isolated and have to be trapped by compounds with a triple or double bond.

Reactions of Salts 1 with Acetylenic Dipolarophiles. In the presence of a slight excess of 1,5-diazabicyclo [4.3.0] non-5-ene (DBN), the 5-aminothiazolium chlorides 1 underwent fast cycloaddition with dimethyl acetylenedicarboxylate (DMAD) at room temperature (Table 1). The reactions afforded the expected pyrroles 4 in good yields and occurred with elimination of *tert*-butyl isothiocyanate as well as benzyl (entries 3, 8, 10) or (ethoxycarbonyl) methyl isothiocyanate (entry 4). These heterocumulenes were easily detected in the reaction medium by ¹H NMR spectroscopy. Starting from 1i (entry 8), PhCH₂NCS was separated from 4f, purified by a bulb-to-bulb distillation under reduced pressure and identified by comparison with a commercial product (IR and NMR data). We assume that these reactions proceed through the mesoionic thiazoles 2 and the well recognized 1,3-dipolar cycloaddition-cycloreversion mechanism. The initially-generated adducts 3 involve the spontaneous extrusion of isothiocyanate R³NCS (Scheme 1).

Under the same conditions, cyclization of 1c, g with methyl propiolate furnished two pyrrole derivatives (Scheme 2). Compound 5 was clearly dominant in these crude mixtures (Table 1, entries 12 and 13). The regioisomers 5 and 6 were separated by a silica gel column chromatography but their structures could

not be fully elucidated in the usual ways, for instance by 13 C NMR spectroscopy (Table 5). NOEDIFF experiments were carried out on samples of **5a** and **6a** in order to perform the regioassignments. Selective irradiation of the dimethylamino group in **6a** (δ 2.55) produces a reproducible enhancement (about 15 %) of the singlet at δ 6.31 attributed to the proton on C-4. On the other hand, irradiation of the NMe₂ in **5a** (δ 2.66) causes no significant perturbation of the signal at δ 6.72. These results allow the assignments described in scheme 2. By analogy, structure **5b** was assigned to the major pyrrole which was obtained from the salt **1g**.

Table 1 - Reactions of 5-Aminothiazolium Salts 1 with DMAD and Methyl	Propiolate in the Presence of
DRN a	-

			P	roducts
entry	educts	solvent	isothiocyanate, R ³	pyrroles (% yield) b
1	1a, E-≡-E	CH ₂ Cl ₂	t-Bu	4a (78)
2	1c, E-≡-E	CH_2Cl_2	t-Bu	4b (61)
3	1d, E-≡-E	CH_2Cl_2	CH_2Ph	4b (81)
4	1e, E-≡-E	CH_2Cl_2	CH ₂ CO ₂ Et	4b (48)
5	1f , E-≡-E	CH ₂ Cl ₂	t-Bu	4c (74)
6	1g, E-≡-E	THF	t-Bu	4d (81)
7	1h, E-≡-E	CH ₂ Cl ₂	t-Bu	4e (60)
8	1i, E-≡-E	CH_2Cl_2	CH_2Ph	4f (57)
9	1j, E-≡-E	CH_2Cl_2	t-Bu	4g (50)
10	1k, E-≡-E	CH ₂ Cl ₂	CH ₂ Ph	4g (64)
11	11, E-≅-E	MeCN	t-Bu	4h (56)
12	1c, E-≡	THF	t-Bu	5a (30); 6a (16) ^c
13	1g, E-≡	THF	t-Bu	5b (32); 6b (14) ^c

^a The reactions were performed at rt for 1 h, starting from a 0.3 M solution of 1 then adding an excess of alkyne (1.5 equiv) and base (1.3 equiv). ^b Purified pyrroles yields after silica gel chromatography and (or) recrystallization. ^c The distributions between the cycloadducts 5, 6 were estimated on the basis on the 1 H NMR spectra of the crude mixtures (5a/6a = 67: 33; 5b/6b = 70: 30).

1c, g
$$\stackrel{\text{i}}{\longrightarrow}$$
 $\stackrel{\text{Me}_2\text{N}}{\longrightarrow}$ $\stackrel{\text{SMe}}{\longrightarrow}$ $\stackrel{\text{SMe}}{\longrightarrow}$ $\stackrel{\text{SMe}}{\longrightarrow}$ $\stackrel{\text{SMe}}{\longrightarrow}$ $\stackrel{\text{Sa, 6a, R}^2 = \text{CH}_2\text{Ph}}{\longrightarrow}$ $\stackrel{\text{Sh, 6b, R}^2 = \text{CH}_2\text{CO}_2\text{Et}}{\longrightarrow}$ $\stackrel{\text{SMe}}{\longrightarrow}$ $\stackrel{\text{Sh, 6b, R}^2 = \text{CH}_2\text{CO}_2\text{Et}}{\longrightarrow}$

Scheme 2 i, DBN, ECECH (- tBuNCS); (E = CO2Me)

Formation of two regioisomeric adducts through 1,3-dipolar cycloaddition reactions with unsymmetrical dipolarophiles has ample precedent in the literature, including cyclic azomethine ylides as münchnones with propiolic esters. The distribution of products depends on the nature and location of the substituents on the starting ylides. 10,11

Reactions of Salts 1 with Olefinic Dipolarophiles. In basic media, the reactions of 1c, g with dimethyl fumarate and maleate proceeded smoothly at room temperature and generally gave a mixture of 1,4-diamino butadiene 7, pyrrole 8 and tetrahydropyrrolo [3,4-b] pyrrole 9. The results were strongly dependent on the nature of the starting reagents and base (Table 2). The reaction with dimethyl fumarate

afforded compounds 8 or 9, according to the presence of NEt3 or DBN (compare entries 4 and 7 with entries 5 and 8). Under identical conditions, the use of methyl maleate gave mostly products 7 or 8 (entries 6, 9). The butadiene 71 was only obtained from the thiazolium chloride 11. In contrast (as shown in entries 1-3), the reactions of salt 1a proceeded with the exclusive formation of the 2-pyrrolines 10a in the presence of NEt3 but gave only 9a in the presence of DBN. Both possible diastereoisomers 10a were detected in the reaction medium and their relative proportions were determined by ¹H NMR spectroscopy (trans/cis ~ 62 : 38 and 77 : 23 using fumarate and maleate, respectively). No epimerization of the isolated pure diastereoisomers 10a occurred under the action of NEt3 at room temperature for about 15 h, indicating the observed stereochemistry to result from kinetic control. On the other hand, we have verified that both isomers 10a underwent fast cyclization to 9a in the presence of DBN ¹², consequently to the expected thermodynamic equilibrium (scheme 3). We also observed two isomers for the butadienes 7c and 7l, one of them being strongly dominant (~ 85 : 15).

Table 2. Reactions of 5-Aminothiazolium Salts 1 with Dimethyl Fumarate and Maleate in Basic Media.

entry	educts	reactn condns a			products distribution b				yields ^c , %
	•	solvent	base	time (h)	7	8	9	10	
1	1a, Fum.	CH ₂ Cl ₂	NEt ₃	13	-	-	-	100	10a (trans) (54); 10a (cis) (32)
2	1a, Mal.	CH ₂ Cl ₂	NEt ₃	14	-	-	-	100	10a (trans) (64); 10a (cis) (12)
3	1a, Fum.	CH ₂ Cl ₂	DBN	2	-	-	100	-	9a (82)
4	1c, Fum.	THF	NEt ₃	3	35	65	-	-	7c (24); 8c (18)
5	1c, Fum.	THF	DBN	0.5	25	-	75	-	7c (19); 9c (44)
6	1c, Mal.	THF	DBN	0.5	75	5	20	-	7c (40)
7	1g, Fum.	THF	NEt ₃	3	-	77	23	-	8g (48)
8	1g, Fum.	THF	DBN	0.5	_	10	90	-	9g (46)
9	1g, Mal.	CH ₂ Cl ₂	DBN	1	_	40	60	-	9g (22); 8g (11)
10	11, Fum.	CH_2Cl_2	DBN	1	100	-	-	-	71 (71)

^a All the reactions were conducted at rt, starting from a 0.3 M solution of 1 then adding the dipolarophile (1.5 - fold excess) and base (1.3 - fold excess). ^b These distributions were calculated on the basis on the ¹H NMR spectra of the crude mixtures. ^c Purified products yields after silica gel chromatography and (or) recrystallization.

All these compounds (except isomers 7) were separated by silica gel column chromatography and were completely characterized by their analytical and spectral data. The shielding of the 4-H and the ester methyl by the *cis*-vic phenyl ¹³ allowed the stereochemical assignment of the diastereoisomers of **10a**: $\delta_{\rm H}$ 4.82 vs 5.20 for 4-H and 3.69 vs 3.21 for 4-CO₂CH₃. The multiplicity of the ¹³C NMR signal attributed to the thione carbon confirmed these assignments: the thioxo carbon appears as a doublet for the *trans*-pyrroline (3 *I*_{CCCH} = 8.5 Hz) and as a singlet for the *cis*-pyrroline, owing to a very low coupling constant (Table 5).

The results in table 2 are rationalized by assuming a typical [3 + 2] cycloaddition of the in situ generated thiazoles 2. The unstable N-bridged cycloadducts 11 do not extrude the tert-butyl isothiocyanate but rather undergo a base-catalyzed rearrangement into the pyrrolines 10 via a 1.5-proton migration from the activated C-6 on the exocyclic nitrogen atom. The formation of pyrroline 10a (trans) from salt 1a and dimethyl fumarate in the presence of NEt3 may be interpreted by the predominant formation of the 5-exo, 6-endointermediate 11a through the transition state described in scheme 4. The minor 10a (cis) arises from the other transition state via the 5-endo, 6-exo-cycloadduct 11a. However, it is clear from the small isomer ratio that the two transition states differ little in energy. The major trans-pyrroline 10a obtained from dimethyl maleate and NEt₃ (which preserves the dipolarophile and cycloadduct geometries) requires the predominant formation of the 5-exo, 6-exo-intermediate 11a and corresponds to an endo-cycloaddition with regard to the dipole. The dihydro pyrroles 10c, g, I cannot be isolated since they are rapidly transformed in basic media according to three different directions: ring opening route (a) to furnish the diaminodienes 7; aromatization step (b) to give the pyrroles 8 via the loss of HNMe2; cyclization process (c) with elimination of MeOH to yield the fused products 9 (Scheme 5). In CHCl₃ solution, some of the bicyclic compounds 9 slowly undergo a new rearrangement to afford the thiomaleimides 12 (1,3-C to N proton shift). The process (c) was much slower under the action of NEt3, as a consequence of which condensed pyrroles 9 were found in lower quantities than in the presence of DBN when we started from 1c, g and dimethyl fumarate. In the case of entries 5, 8 the formation of compounds 9 could be interpreted by a suitable structure cis for the precursors 10 (Scheme 5). This stereochemistry results from the corresponding 5-endo, 6-exo-intermediates 11. We presume that the use of maleate leads predominantly to the trans-pyrrolines 10, before isomerisation. This stereochemistry requires an endo-transition state with regard to the dipole and the formation of the transient 5-exo, 6-exo-cycloadducts 11.

Treatment of thiazolium chlorides 1 with fumaronitrile afforded similar results. The cycloadditions proceeded rapidly at room temperature to give predominantly one of the following compounds according to the nature of the starting salt and base: diaminobutadiene 14, pyrrole 15 and fused pyrroline 16 (Table 3). The 2-pyrrolines 13a, b were obtained in small quantities (together with condensed adducts 16a, b) from the salts 1a,b under the action of NEt₃ (a single isomer 13 with an undetermined *cis* or *trans*-structure). The reactions were probably proceeding under thermodynamic control. The results of table 3 can be explained by the 1,3-dipolar trapping of mesoionic thiazoles 2 followed by base-catalyzed rearrangements of the initially-formed cycloadducts (Scheme 6).

Table 3. Reactions of 5-Aminothiazolium Salts 1 with Fumaronitrile in Basic Media (rt, 1 h)a

salt	reactn c	reactn condns products distr		stribution	₁ b	yields ^c , %	
	solvent	base	13	14	15	16	
1a	CH ₂ Cl ₂	NEt ₃	35	-	-	65	13a (32); 16a (61)
1a	CH_2Cl_2	DBN	-	-	-	100	16a (89)
1 b	CH_2Cl_2	NEt ₃	20	-	-	80	13b (17); 16b (64)
1 b	CH_2Cl_2	DBN	-	-	-	100	16b (88)
1 c	THF	NEt_3	-	90	2	8	14c (55)
1 c	THF	DBN	-	17	83	-	15c (50)
1 f	CH_2Cl_2	DBN	-	22	78	-	15f (48)
1 g	THF	DBN	-	-	100	-	15g (57)

a,b,c: see Table 2.

A few examples of analogous base-induced rearrangements, without any fragmentation, have been reported in the literature, starting from standard electron-deficient olefins and cyclic azomethine ylides as imidazolium-4-olates, ¹⁴ munchnones, ¹⁵ azlactones ¹⁶ and Reissert salts. ¹⁷ [3 + 2] Cycloaddition reactions of oxazolium-5-olates generally give 2-pyrrolines via usual CO₂ elimination. ¹⁸

Reactions of Salts 1 with N-phenylmaleimide. The reactivity of mesoionic compounds such as 2a, c, j as 4π electron-rich system was also demonstrated by their fast 1,3-dipolar cycloaddition with N-phenylmaleimide. Salt 1a readily underwent addition in the presence of NEt3 to afford a product which was identified as the 1:1 primary adduct 17a. The latter was found to be stable in the solid state for several months but was rapidly degraded by DBN to give unidentified compounds. The stereochemical assignment of 17a was based only on NMR method: the *cis*-protons 3a-H and 7a-H couple with 9.7 Hz, consistent with the experience on structurally related *endo*-adducts (3J = 8.4-9.6 Hz) while the *exo*-isomers show 6.7-6.8 Hz, 19-21

Subjected to the same conditions, salts 1c, j produced two succinimide derivatives 19, 20 in moderate yields (Table 4). Attempts to isolate the primary adducts 17 were fruitless. Compounds 19 and 20 may be the result of a base-induced rearrangement, without the further loss of *tert*-butyl isothiocyanate, as described in scheme 7. In particular, we postulate that the conversion of the unstable species 18 can occur in two ways: ring opening of the six-membered cycle (a) or elimination of MeSH (b).

Table 4. Reactions of Salts 1 with N-Phenylmaleimide in Basic Media (CH₂Cl₂, rt. 1 h)^a

salt	base	products distribution b		oution ^b	yields ^c , %
		17	19	20	-
1a	NEt ₃	100	-	-	17a (83)
1 c	NEt ₃	-	35	65	19c (29) ; 20c (44)
1 c	DBN	-	-	100	20c (54)
1 j	NEt ₃	-	25	75	20j (40)
1 j	DBN	-	-	100	20j (46)

a,b,c: see Table 2.

In neutral media, stable initial 1:1 cycloadducts are frequently isolated from cyclic azomethine ylides and olefinic dipolarophiles.³ As an example, mesoionic 1,3-thiazol-5-ones readily give such adducts with N-phenylmaleimide ²⁶ and maleic anhydride.²⁷ Their endo or exo stereochemical features have been discussed.

Conclusion. We have shown that the 5-aminothiazolium salts 1 react easily as cyclic azomethine ylides in basic media. 1,3-Dipolar cycloadditions of the corresponding mesoionic thiazoles 2 occur under fairly mild conditions and open up synthetic possibilities for a variety of functionalized monocyclic or condensed pyrrolines and pyrroles. Numerous examples are given and mechanisms have been suggested to explain the observed rearrangements. Further studies on these systems deserve some attention and other examples using heterocumulenes as dipolarophiles will be published shortly.

Experimental section

General. Melting points are uncorrected. ¹H NMR spectra (80 or 300 MHz) and ¹³C NMR spectra (75.5 MHz) were recorded in CDCl₃. When necessary, unambiguous NMR assignments were acquired by decoupling experiments. HRMS were obtained from the Centre Régional de Mesures Physiques de l'Ouest, in the electron impact mode, using a potential of 70 eV. With the exception of molecular ion peaks, only mass spectral fragments with relative intensities of 15 % or more are reported. Infrared spectra were recorded as suspensions in Nujol. Elemental analyses were performed by the analytical laboratory, CNRS.

The following reactions were conducted under a dry nitrogen atmosphere. Na₂SO₄ was used to dry organic layers after extractions. Crude products were generally fractionated by silica gel column flash

chromatography then purified by recrystallization. Corresponding eluents and solvents will be given below with analytical and spectral data of various compounds.

Reactions of Salts 1 with Acetylenic Dipolarophiles. Preparation of Pyrroles 4, 5, 6. DBN (0.97 g, 7.8 mmol) was added dropwise to a mixture of 1 (6 mmol) and DMAD (1.28 g, 9 mmol) or methyl propiolate (0.76 g, 9 mmol) in anhydrous CH₂Cl₂, THF or MeCN (20 mL, Table 1). The solution was maintained at rt for 1 h then the solvent was removed under reduced pressure. The residue was poured into 30 mL of Et₂O and washed with H₂O. The ¹H NMR analysis of the crude oily product showed the formation of pyrrole and isothiocyanate in the ratio 50: 50. These compounds were separated by chromatography except for 4d and 4h which precipitated by trituration with petroleum ether (yields and ¹³C NMR spectra, see Tables 1 and 5; 4b, see our preceding article ²).

Dimethyl 1-Methyl-5-phenyl-2-(phenylthio)-1H-pyrrole-3,4-dicarboxylate (4a): Et₂O/petroleum ether (1:1) as eluent; mp 103°C (MeOH); 1 H NMR δ 3.27 (s, 3H), 3.56 (s, 3H), 3.78 (s, 3H), 7.30 (m, 10H); MS calcd for C₂₁H₁₉NO₄S m/z 381.1035 (M⁺), found 381.1046; m/z (rel int) 381 (100), 318 (30); IR 1703 cm⁻¹. Anal. Calcd for C₂₁H₁₉NO₄S: C, 66.14; H, 4.99; N, 3.67; S, 8.40. Found: C, 66.44; H, 5.00; N, 3.65; S, 8.06.

Dimethyl 1-Benzyl-5-(dimethylamino)-2-(phenylthio)-1H- pyrrole-3,4-dicarboxylate (4c): Et₂O/petroleum ether (1:1) as eluent; mp 111°C (MeOH); ¹H NMR δ 2.61 (s, 6H), 3.81 (s, 6H), 5.12 (s, 2H), 7.05 (m, 10H); MS calcd for $C_{23}H_{24}N_{2}O_{4}S$ m/z 424.1457 (M⁺), found 424.1462; m/z (rel int) 424 (30), 333 (100), 269 (27); IR 1700 cm⁻¹. Anal. Calcd for $C_{23}H_{24}N_{2}O_{4}S$: C, 65.09; H, 5.66; N, 6.60; S, 7.55. Found: C, 65.44; H, 5.95; N, 6.73; S, 7.63.

Dimethyl 5-(Dimethylamino)-1-[(ethoxycarbonyl)methyl]-2-(methylthio)-1H-pyrrole-3,4-dicarboxylate (4d): mp 62°C (ether/petroleum ether); 1 H NMR δ 1.22 (t, J = 7 Hz, 3H), 2.26 (s, 3H), 2.67 (s, 6H), 3.77 (s, 3H), 3.83 (s, 3H), 4.18 (q, J = 7 Hz, 2H), 4.80 (s, 2H); MS calcd for $C_{15}H_{22}N_{2}O_{6}S$ m/z 358.1198 (M⁺), found 358.1174; m/z (rel int) 358 (83), 343 (100), 327 (15), 239 (16), 194 (24); IR 1740, 1715, 1700 cm⁻¹. Anal. Calcd for $C_{15}H_{22}N_{2}O_{6}S$: C, 50.28; H, 6.14; N, 7.82; S, 8.94. Found: C, 50.34; H, 6.21; N, 7.72; S, 8.80.

Dimethyl 5-(Dimethylamino)-1-[(ethoxycarbonyl)methyl]-2-(phenylthio)-1H- pyrrole-3,4-dicarboxylate (4e): Et₂O/petroleum ether (1:1) as eluent; bp 180°C (0.02 Torr) (Buchi Kugelrohr apparatus); 1 H NMR δ 1.08 (t, J = 7 Hz, 3H), 2.71 (s, 6H), 3.81 (s, 6H), 3.95 (q, J = 7 Hz, 2H), 4.66 (s, 2H), 7.18 (s, 5H); MS calcd for C₂₀H₂₄N₂O₆S m/z 420.1355 (M[±]), found 420.1375; m/z (rel int) 420 (100), 333 (26). Anal. Calcd for C₂₀H₂₄N₂O₆S: C, 57.14; H, 5.71; N, 6.67; S, 7.62. Found: C, 56.95; H, 5.69; N, 6.39; S, 7.79.

Dimethyl 1-[(Diethoxyphosphoryl)methyl]-5-(dimethylamino)-2-(methylthio)-1H-pyrrole-3,4-dicarboxylate (4f): Et₂O as eluent; crude oil; ¹H NMR δ 1.31 (t, J = 7 Hz, 6H), 2.40 (s, 3H), 2.77 (s, 6H), 3.80 (s, 3H), 3.85 (s, 3H), 4.11 (m, 4H), 4.56 (d, ${}^2J_{\text{HCP}}$ = 16 Hz, 2H); MS calcd for C₁₆H₂₇N₂O₇PS m/z 422.1276 (M⁺), found 422.1272; m/z (rel int) 422 (100), 407 (74), 375 (27), 271 (15), 149 (26).

Dimethyl 5-(Dimethylamino)-2-(methylthio)-1-isopropyl-1H- pyrrole-3,4-dicarboxylate (4g): Et₂O/petroleum ether (1:2) as eluent; mp 90°C (MeOH); 1 H NMR δ 1.56 (d, J = 7 Hz, 6H), 2.33 (s, 3H), 2.76 (s, 6H), 3.78 (s, 3H), 3.85 (s, 3H), 5.08 (m, 1H); MS calcd for $C_{14}H_{22}N_{2}O_{4}S$ m/z 314.1300 (M⁺), found 314.1322; m/z (rel int) 314 (55), 271 (100), 239 (47), 207 (16). Anal. Calcd for $C_{14}H_{22}N_{2}O_{4}S$: C, 53.50; H, 7.01; N, 8.92; S, 10.19. Found: C, 53.17; H, 7.06; N, 8.80; S, 10.08.

Dimethyl 5-(Dimethylamino)-1-(2,6-dimethylphenyl)-2-(methylthio)-1H- pyrrole-3,4-dicarboxylate (4h): mp 97°C (MeOH); 1 H NMR δ 1.96 (s, 6H), 2.03 (s, 3H), 3.81 (s, 6H), 3.79 (s, 3H), 3.85 (s, 3H), 7.17 (m, 3H); MS calcd for $C_{19}H_{24}N_2O_4S$ m/z 376.1457 (M⁺), found 376.1445; m/z (rel int) 376 (92), 361 (100), 329 (20). Anal. Calcd for $C_{19}H_{24}N_2O_4S$: C, 60.64; H, 6.38; N, 7.45; S, 8.51. Found: C, 60.35; H, 6.77; N, 7.21; S, 8.50.

 $\label{eq:methylamino} \begin{tabular}{ll} \begin{tabular}{ll} Methyl & 1-Benzyl-5-(dimethylamino)-2-(methylthio)-1H-pyrrole-4-carboxylate & (5a) \\ \hline $:$ Et_2O/petroleum ether (1:2) as eluent; bp 160°C (0.02 Torr) (Buchi); 1H NMR $$\delta$ 2.01 (s, 3H), 2.66 (s, 6H), 3.77 (s, 3H), 5.24 (s, 2H), 6.72 (s, 1H), 7.10 (m, 5H); MS calcd for $C_{16}H_{20}N_2O_2S$ m/z 304.1245 (M$^+$), found 304.1257; m/z (rel int) 304 (42), 213 (100), 181 (26), 91 (20); IR 1700 cm$^-1. Anal. Calcd for $C_{16}H_{20}N_2O_2S$: C, 63.16; H, 6.58; N, 9.21. Found : C, 62.78; H, 6.73; N, 9.01. \end{tabular}$

 $\label{eq:methylamino} \begin{tabular}{ll} Methyl & 1-Benzyl-5-(dimethylamino)-2-(methylthio)-1H-pyrrole-3-carboxylate & (6a) \\ : mp & 68°C & (CH_2Cl_2/Et_2O) ; $^1H NMR & 2.12 & (s, 3H), 2.55 & (s, 6H), 3.80 & (s, 3H), 5.31 & (s, 2H), 6.31 & (s, 1H), \\ 7.10 & (m, 5H) ; MS & found & 304.1248 & ; m/z & (rel int) & 304 & (22), 213 & (100), 181 & (53), 91 & (27) & ; IR & 1720 & cm^{-1}. \\ Anal. & Found & : C, & 62.83 & ; H, & 6.57 & ; N, 9.19. \\ \end{tabular}$

Methyl 5-(Dimethylamino)-1-[(ethoxycarbonyl)methyl]-2-(methylthio)-1H- pyrrole-4-carboxylate (5b): Et₂O/petroleum ether (1:2) as eluent; mp 49°C (MeOH); 1 H NMR δ 1.24 (t, J = 7 Hz, 3H), 2.17 (s, 3H), 2.71 (s, 6H), 3.75 (s, 3H), 4.20 (q, J = 7 Hz, 2H), 4.78 (s, 2H), 6.72 (s, 1H); MS calcd for $C_{13}H_{20}N_{2}O_{4}S$ m/z 300.1144 (M $^{+}$), found 300.1146; m/z (rel int) 300 (80), 285 (100); IR 1745, 1705 cm $^{-1}$. Anal. Calcd for $C_{13}H_{20}N_{2}O_{4}S$: C, 52.00; H, 6.67; N, 9.33; S, 10.67. Found: C, 51.70; H, 6.60; N, 9.26; S, 10.37.

Methyl 5-(Dimethylamino)-1-[(ethoxycarbonyl)methyl]-2-(methylthio)-1H-pyrrole-3-carboxylate (6b): bp 160° C (0.02 Torr); ¹H NMR δ 1.26 (t, J=7 Hz, 3H), 2.31 (s, 3H), 2.57 (s, 6H), 3.80 (s, 3H), 4.21 (q, J=7 Hz, 2H), 4.81 (s, 2H), 6.24 (s, 1H); MS found 300.1136; m/z (rel int) 300 (53), 285 (100); IR 1740, 1702 cm⁻¹. Anal. Found: C, 51.63; H, 6.61; N, 9.04.

Reactions of Salts 1 with Olefinic Dipolarophiles. A solution of 1 (6 mmol) in dry CH_2Cl_2 or THF (30 mL) was prepared. Dimethyl fumarate, maleate, fumaronitrile or N-phenylmaleimide (9 mmol) then NEt₃ or DBN (7.8 mmol) were added dropwise and the mixture was stirred at rt for the time indicated in Tables 2, 3 and 4. As precedently described, workup procedure consists of concentration, addition of water, extraction with ether and flash chromatographic separation of the various products (yields and ¹³C NMR data, see Tables 2-6). Compounds **14f**, **16c** and **19j** were not thoroughly purified and were only characterized by ¹H NMR. [**14f** : δ 1.61 (s, 9H), 2.77 (d, 2H), 3.13 (s, 6H), 5.74 (t, br, 1H), 7.25 (m, 10H), 8.42 (br, 1H). **16c** : δ 1.82 (s, 9H), 2.18 (s, 6H), 2.63 (s, 3H), 3.78 (s, 1H), 4.47, 5.35 (AB syst, J = 17.6 Hz, 2H), 7.25 (m, 5H). **19j** : δ 1.25 (d, J = 6 Hz, 6H), 1.38 (s, 9H), 2.26 (s, 3H), 2.77 (s, 3H), 3.01 (s, 3H), 4.07 (m, 1H), 6.86 (br, 1H), 7.30 (s, 5H), 8.55 (d, J = 9 Hz, 1H)].

Dimethyl 1-(Benzylamino)-4-(*tert*-butylthiocarbamoyle)-4-(dimethylamino)-1-(methylthio) butadiene-2,3-dicarboxylate (7c): Et₂O/petroleum ether (6:1) as eluent; mp 125 °C (Et₂O); ¹H NMR (prevailing isomer) δ 1.38 (s, 9H), 2.23 (s, 3H), 2.95 (s, 6H), 3.62 (s, 6H), 4.67 (m, 2H), 7.27 (s, 5H), 7.62 (br, 1H), 9.31 (t, br, J = 6 Hz, 1H); ¹³C NMR (prevailing isomer) δ 18.5 (q, ¹J = 141 Hz), 27.0 (qm, ¹J = 127 Hz), 43.1 (qq, ¹J = 140 Hz, ³J = 3.6 Hz), 49.8 (tm, ¹J = 137 Hz), 51.0, 51.1 (2q, ¹J = 145 Hz), 55.6 (m), 96.4, 101.2 (2s), 127.2, 127.3 (2 dm, ¹J = 160 Hz), 128.6 (dd, ¹J = 160 Hz, ³J = 6 Hz), 138.9, 161.4, 165.9 (3 m), 168.1, 170.2 (2 q, ³J = 3.7 Hz), 194.4 (s); MS calcd for C₂₃H₃₃N₃O₄S₂ m/z 479.1912 (M[±]), found 479.1918; m/z (rel int) 479 (1), 348 (17), 91 (100); IR 3322, 1670, 1638 cm⁻¹. Anal.

Calcd for $C_{23}H_{33}N_3O_4S_2$: C, 57.62; H, 6.89; N, 8.77; S, 13.36. Found: C, 57.43; H, 6.97; N, 8.44; S, 13.69.

Dimethyl 4-(*tert*- Butylthiocarbamoyle)-4-(dimethylamino)-1-[(2,6-dimethylphenyl) amino]-1-(methylthio) butadiene-2,3-dicarboxylate (7l): Et₂O/CH₂Cl₂ (1:2) as eluent; mp 173°C (MeOH); ¹H NMR (prevailing isomer) δ 1.46 (s, 9H), 1.71 (s, 3H), 2.20 (s, 3H), 2.30 (s, 3H), 2.93 (s, 6H), 3.62 (s, 3H), 3.66 (s, 3H), 7.02 (s, 3H), 7.75 (br, 1H), 10.33 (br, 1H); ¹³C NMR (prevailing isomer) δ 16.2 (q, ¹J = 142 Hz), 18.6, 19.1 (2 qd, ¹J = 127 Hz, ³J = 3 Hz), 27.1 (qm, ¹J = 127 Hz), 43.1 (qq, ¹J = 138 Hz, ³J = 3 Hz), 51.1, 51.2 (2 q, ¹J = 146 Hz), 55.6 (m), 97.5, 101.2 (2s), 126.6 (d, ¹J = 160 Hz), 128.1, 128.5 (2 dm, ¹J = 158 Hz), 135.9, 138.0, 160.7, 165.5 (4 m), 168.3, 170.1 (2 q, ³J = 3.6 Hz), 194.6 (s); MS calcd for C₂₄H₃₅N₃O₄S₂ m/z 493.2069 (M⁺), found 493.2067; m/z (rel int) 493 (61), 446 (20), 377 (17), 362 (17), 350 (15), 345 (21), 315 (54), 314 (48), 295 (49), 283 (21), 178 (84), 91 (100); IR 3300, 1660, 1630 cm⁻¹. Anal. Calcd for C₂₄H₃₅N₃O₄S₂: C, 58.42; H, 7.10; N, 8.52; S, 12.98. Found: C, 58.29; H, 7.00; N, 8.47; S, 12.91.

Dimethyl 1-Benzyl-5-(*tert*-butylthiocarbamoyl)-2-(methylthio)-1H- pyrrole-3,4-dicarboxylate (8c): Et₂O/petroleum ether (1:1) as eluent; mp 135°C (MeOH); 1H NMR δ 1.33 (s, 9H), 2.30 (s, 3H), 3.76 (s, 3H), 3.87 (s, 3H), 5.75 (s, 2H), 7.10 (m, 5H); MS calcd for $C_{21}H_{26}N_{2}O_{4}S_{2}$ m/z 434.1334 (M⁺), found 434.1339; m/z (rel int) 434 (10), 377 (33), 223 (16), 91 (100); IR 3260, 1700 cm⁻¹. Anal. Calcd for $C_{21}H_{26}N_{2}O_{4}S_{2}$: C, 58.06; H, 5.99; N, 6.45; S, 14.75. Found: C, 58.30; H, 5.84; N, 6.29; S, 14.63.

Dimethyl 5-(tert-Butylthiocarbamoyl)-1-[(ethoxycarbonyl)methyl]-2-(methylthio)-1H-pyrrole-3,4-dicarboxylate (8g): Et₂O/petroleum ether (1:1) as eluent; mp 138°C (MeOH); 1 H NMR δ 1.27 (t, J = 7 Hz, 3H); 1.56 (s, 9H), 2.31 (s, 3H), 3.77 (s, 3H), 3.84 (s, 3H), 4.20 (q, J = 7 Hz, 2H), 5.31 (s, 2H), 7.86 (br, 1H); MS calcd for $C_{18}H_{26}N_{2}O_{6}S_{2}$ m/z 430.1232 (M⁺), found 430.1250; m/z (rel int) 430 (69), 401 (27), 341 (37), 313 (24), 309 (100); IR 3280, 1745, 1705 cm⁻¹. Anal. Calcd for $C_{18}H_{26}N_{2}O_{6}S_{2}$: C, 50.23; H, 6.05; N, 6.51; S, 14.88. Found: C, 50.11; H, 6.01; N, 6.47; S, 14.83.

Methyl 5-tert-Butyl-1-methyl-4-oxo-6a-phenyl-2-(phenylthio)-6-thioxo-3a,4,5,6a-tetrahydro-1H, 6H-pyrrolo [3,4-b] pyrrole-3-carboxylate (9a): Et₂O/petroleum ether (1:1) as eluent; mp 131°C (MeOH); $^1\mathrm{H}$ NMR δ 1.78 (s, 9H), 2.80 (s, 3H), 3.65 (s, 3H), 3.98 (s, 1H), 7.20 (m, 10H); MS calcd for $C_{25}H_{26}N_{2}O_{3}S_{2}$ m/z 466.1385 (M⁺), found 466.1372; m/z (rel int) 466 (100), 410 (27), 377 (15), 350 (21), 323 (89); IR 1745, 1680 cm⁻¹. Anal. Calcd for $C_{25}H_{26}N_{2}O_{3}S_{2}$: C, 64.38; H, 5.58; N, 6.01. Found: C, 64.30; H, 5.57; N, 6.07.

Methyl 1-Benzyl-5-tert-butyl-6a-(dimethylamino)-2-(methylthio)-4-oxo-6-thioxo-3a,4,5,6a-tetrahydro-1H, 6H-pyrrolo [3,4-b] pyrrole-3-carboxylate (9c): Et₂O/petroleum ether (2:1) as eluent; mp 128°C (MeOH); 1 H NMR δ 1.66 (s, 9H), 2.16 (s, 6H), 2.41 (s, 3H), 3.81 (s, 3H), 4.06 (s, 1H), 4.70, 5.31 (AB syst, J=17.4 Hz, 2H), 7.08 (m, 5H); MS calcd for $C_{22}H_{29}N_{3}O_{3}S_{2}$ m/z 447.1650 (M⁺), found 447.1653; m/z (rel int) 447 (11), 300 (17), 91 (100); IR 1750, 1670 cm⁻¹. Anal. Calcd for $C_{22}H_{29}N_{3}O_{3}S_{2}$: C, 59.06; H, 6.49; N, 9.40; S, 14.32. Found: C, 59.13; H, 6.65; N, 9.27; S, 14.73.

Methyl 5-tert-Butyl-6a-(dimethylamino)-1-[(ethoxycarbonyl) methyl]-2-(methylthio)-4-oxo-6-thioxo-3a,4,5,6a-tetrahydro-1H, 6H- pyrrolo [3,4-b] pyrrole-3-carboxylate (9g): Et₂O/petroleum ether (2:1) as eluent; mp 100° C (CH₂Cl₂/Et₂O); ¹H NMR δ 1.20 (t, J = 7 Hz, 3H), 1.70 (s, 9H), 2.11 (s, 6H), 2.45 (s, 3H), 3.77 (s, 3H), 3.98 (s, 1H), 4.10 (m, 2H), 4.63, 4.25 (AB syst, J = 18 Hz, 2H); MS calcd for C₁9H₂9N₃O₅S₂ m/z 443.1549 (M⁺), found 443.1582; m/z (rel int) 443 (26), 428

(77), 372 (100), 340 (50), 327 (23), 300 (32), 285 (29); IR 1735, 1665 cm⁻¹. Anal. Calcd for $C_{19}H_{29}N_3O_5S_2$: C, 51.47; H, 6.55; N, 9.48; S, 14.45. Found: C, 51.44; H, 6.80; N, 9.32; S, 14.29.

Dimethyl 5-(tert-Butylthiocarbamoyle)-1-methyl-5-phenyl-2-(phenylthio)-4,5-dihydro-1H-pyrrole-3,4-dicarboxylate (10a): Et₂O/petroleum ether (1:1) as eluent; *trans*-isomer: mp 179°C (MeOH); ^1H NMR δ 1.50 (s, 9H), 2.46 (s, 3H), 3.21 (s, 3H), 3.63 (s, 3H), 5.20 (s, 1H), 7.30 (m, 10H), 8.01 (br, 1H); MS calcd for C₂₆H₃₀N₂O₄S₂ m/z 498.1647 (M⁺), found 498.1647; m/z (rel int) 498 (1), 382 (94), 381 (82), 350 (100), 318 (39), 292 (24), 273 (52); IR 3325, 1741, 1698 cm⁻¹. Anal. Calcd for C₂₆H₃₀N₂O₄S₂: C, 62.65; H, 6.02; N, 5.62; S, 12.85. Found: C, 62.64; H, 6.11; N, 5.62; S, 12.47; cis-isomer: mp 133°C (MeOH); ^1H NMR δ 1.44 (s, 9H), 2.13 (s, 3H), 3.65 (s, 3H), 3.69 (s, 3H), 4.82 (s, 1H), 7.30 (m, 10H), 8.10 (br, 1H); MS calcd for C₂₅H₂₆N₂O₃S₂ m/z 466.1385 (M-MeOH)⁺, found 466.1379; m/z (rel int) 466 (89), 410 (33), 389 (16), 350 (22), 323 (100); IR 3275, 3250, 1720, 1687 cm⁻¹. Anal. Found: C, 62.58; H, 6.01; N, 5.65; S, 12.94.

4-[β-(Benzylamino)-α-(methoxycarbonyl)-β-(methylthio) vinyl]-1-tert-butyl-3-(dimethylamino)-2-thiomaleimide (12c): mp 98°C (MeOH); 1 H NMR δ 1.73 (s, 9H), 2.25 (s, 3H), 2.93 (s, 6H), 3.62 (s, 3H), 4.70 (d, J=6 Hz, 2H), 7.30 (s, 5H), 9.66 (br, 1H); 13 C NMR δ 17.8 (q, $^{1}J=141$ Hz), 24.4 (qm, $^{1}J=128$ Hz), 42.7 (qq, $^{1}J=137$ Hz, $^{3}J=3.9$ Hz), 49.7 (tm, $^{1}J=138$ Hz), 51.2 (q, $^{1}J=146$ Hz), 60.4 (m), 91.8 (d, $^{3}J=2$ Hz), 106.6 (s), 127.0 (dm, $^{1}J=158$ Hz), 127.4 (dt, $^{1}J=161$ Hz, $^{3}J=6.8$ Hz), 128.7 (dd, $^{1}J=160$ Hz, $^{3}J=6.8$ Hz), 138.6, 152.7, 164.0 (3 m), 169.2 (q, $^{3}J=3.8$ Hz), 174.9, 199.3 (3s); MS calcd for C₂₂H₂₉N₃O₃S₂ m/z 447.1650 (M†-), found 447.1678; m/z (rel int) 447 (29), 91 (100); IR 3180, 1717, 1640 cm⁻¹. Anal. Calcd for C₂₂H₂₉N₃O₃S₂: C, 59.06; H, 6.49; N, 9.40. Found: 58.66; H, 6.69; N, 9.12.

1-tert-Butyl-3-(dimethylamino)-4-[β-[[(ethoxycarbonyl)methyl] amino]-α-(methoxycarbonyl)-β-(methylthio) vinyl]-2-thiomaleimide (12g): mp 209°C (MeOH); 1 H NMR δ 1.24 (t, J=7 Hz, 3 H), 1.49 (s, 9 H), 2.36 (s, 3 H), 3.27 (s, 6 CH), 3.68 (s, 3 H), 4.17 (q, J=7 Hz, 2 CH), 4.57 (s, 3 CH), 7.70 (br, 3 H); 3 C NMR δ 14.2 (qt, 3 J=127 Hz, 3 J=3.0 Hz), 20.6 (q, 3 J=142 Hz), 27.0 (qm, 3 J=127 Hz), 42.6 (t, 3 J=141 Hz), 45.1 (qbr, 3 J=137 Hz), 50.8 (q, 3 J=147 Hz), 56.7 (m), 61.4 (tq, 3 J=148 Hz, 2J=4.5 Hz), 97.8, 114.9 (2s), 133.3, 161.6 (2m), 162.7 (s), 164.3 (q, 3 J=4.0 Hz), 169.3 (m), 190.0 (s); MS calcd for 3 C₁9H₂9N₃O₅S₂ m/z 443.1549 (M⁺), found 443.1539; m/z (rel int) 443 (100), 428 (22), 386 (15), 354 (22), 340 (20), 338 (16), 315 (17), 313 (19); IR 3240, 1740, 1705 cm⁻¹. Anal. Calcd for 3 C₁9H₂9N₃O₅S₂: C, 51.47; H, 6.55; N, 9.48. Found: C, 51.07; H, 6.65; N, 9.09.

 $\begin{array}{c} \textbf{5-(tert-Butylthiocarbamoyle)-1-methyl-5-phenyl-2-(phenylthio)-4,5-dihydro-1H-pyrrole-3,4-dicarbonitrile (13a):} & Et_2O/petroleum ether (4:1) as eluent; mp 130°C (MeOH); 1H NMR $ 1.36 (s, 9H), 2.70 (s, 3H), 6.36 (s, 1H), 7.30 (m, 10H); MS calcd for $C_{24}H_{24}N_4S_2$ m/z 432.1442 (M$^+$), found 432.1433; m/z (rel int) 432 (8), 315 (100); IR 3280, 2240, 2185 cm$^-1$. Anal. Calcd for $C_{24}H_{24}N_4S_2$: $C, 66.67; H, 5.56; N, 12.96; S, 14.81. Found: $C, 66.50; H, 5.53; N, 12.80; S, 14.50. \end{array}$

 $\begin{array}{c} \textbf{5-(tert\text{-}Butylthiocarbamoyle)\text{-}1\text{-}methyl\text{-}2\text{-}[(4\text{-}methylphenyl)oxy]\text{-}5\text{-}phenyl\text{-}4,5\text{-}} \\ \textbf{dihydro\text{-}1H\text{-}pyrrole\text{-}3,4\text{-}dicarbonitrile} & (13b) : Et_2O/\text{petroleum} & \text{ether} & (3:2) \text{ as eluent} ; mp 210°C \\ (\text{MeOH}) ; ^1\text{H} & \text{NMR} & 1.38 & (s, 9H), 2.26 & (s, 3H), 2.76 & (s, 3H), 6.01 & (s, 1H), 7.25 & (m, 9H) ; MS calcd for \\ \textbf{C}_{25}\text{H}_{26}\text{N}_{4}\text{OS} & \text{m/z} & 430.1827 & (\text{M}^{+}), \text{ found} & 430.1856 ; \text{m/z} & (\text{rel int)} & 430 & (1), 313 & (100) ; IR 3270, 2238, 2181 \\ \text{cm}^{-1}. & \text{Anal. Calcd for C}_{25}\text{H}_{26}\text{N}_{4}\text{OS} : C, 69.77 ; H, 6.05 ; N, 13.02 ; S, 7.44. Found : C, 69.61 ; H, 6.08 ; N, 12.73 ; S, 7.14. \\ \end{array}$

Table 5. Selected 13 C NMR Chemical Shifts at 75.469 MHz for the Substituted Pyrroles 4, 5, 6, 8, 15 and Dihydro Pyrroles 10, 13 (endocyclic and possible thione carbons)^a, Mult (J, Hz).

n°	C-2	C-3	C-4	C-5 (m)	C=S
4a	121.4 q (3.7)b	125.0 s	113.5 s	140.5	-
4 c	115.9 t (4.3) b	125.3 s	108.2 s	148.3	-
4d	121.6 m	122.5 s	108.4 s	146.9	-
4 e	116.0 t (4.0) ^b	125.2 s	108.2 s	148.3	-
4f	122.2 m	122.4 s	109.1 s	146.7	-
4 g	119.6 m	123.5 s	108.5 s	147.3	-
4 h	121.4 q (4.6) ^c	122.1 s	106.1 s	147.2	-
5a	119.8 m	116.8 d (177.0) ^d	108.3 d (2.2) e	148.6	-
6a	125.2 m	117.9 d (2.2) e	99.1 d (174.5) ^d	146.2	-
5 b	119.5 m	117.5 d (177.7) ^d	108.5 d (2.2) e	148.6	-
6 b	125.1 m	118.2 d (2.1) ^e	98.9 d (175.0) ^d	145.9	-
8 c	129.1 q (4.8) ^c	112.7 s	121.6 s	138.5	186.4 s
8 g	129.5 q (4.4) ^c	112.9 s	121.6 s	138.4	185.9 s
15c	137.8 q (4.8) ^C	94.6 s	100.9 s	142.9	182.4 s
15f	133.8 t (4.2) b	94.8 s	103.8 s	143.5	182.2 s
15g	138.3 q (5.0) ^c	94.5 s	101.9 s	142.8	181.8 s
10a (trans)	152.9 m	101.9 d (5.9) e	60.4 d (140.0) ^d	84.5	199.0 d (8.5) ^f
10a (cis)	155.2 m	109.7 d (4.9) e	60.7 d (146.5) ^d	85.7	195.8 s
13a	156.2 m	80.6 d (7.5) e	46.6 d (145.5) ^d	85.0	194.2 d (8.5) ^f
13b	164.3 m	54.4 d (7.0) e	44.9 d (146.7) ^d	81.0	195.1 d (7.5) ^f

^a The ring carbons are numbered in the way to have the methylthio, phenylthio or aryloxy group on the C-2. ^b $^3J_{(CNCH)}$. ^c $^3J_{(CSCH)}$. ^d $^1J_{.}$ ^e $^2J_{(CCCH)}$. ^f $^3J_{(CCCH)}$.

Table 6. Selected 13 C NMR Chemical Shifts at 75.469 MHz for the Fused Pyrroles **9**, **16** and Thiopyrans **17**, **20**, Mult (J, Hz).

n°	C -1	C-2 (m)	C-3	C-3a	C-4	C-6	C-6a (m)	C-7 (m)	C-7a
9a	-	154.2	99.3 d	58.4 d	176.4 d	206.9 d	80.2	-	-
			$(6.0)^{a}$	(148.6) b	$(3.7)^{a}$	(3.8) ^c			
9 c	-	160.8	96.7 d	43.3 d	177.3 d	204.7 d	90.4	-	-
			$(6.0)^{a}$	(142.6) b	$(5.5)^{a}$	(1.5) ^c			
9 g	-	159.4	97.9 d	42.9 d	176.7 d	204.7 d	89.3	-	-
			$(6.2)^{a}$	(143.3) b	$(5.3)^{a}$	(1.9) ^c			
16a	-	158.0	78.2 d	58.5 d	168.9 br	204.0 d	82.3	-	-
			$(4.9)^{a}$	(148.0) b		(6.0) ^c			
16b	-	165.3	52.5 d	55.9 d	169.9 br	204.6 d	78.2	-	-
			$(4.6)^{a}$	(147.9) b		(4.9) ^c			
17a	172.0 t	-	172.3 t	51.9 dd	92.0 m	155.5 d	•	81.5	53.9 do
	(3.4) a,c		$(4.7)^{a,c}$	(146.6 ^b ,		(5.9) ^c			(147.0 ^t
				3.3 ^a)					3.3 ^a)
20c	164.3 s	-	166.0 s	91.4 s	152.1 t	140.1 s	-	138.4	113.5
					(4.3) d				
20j	164.3 s	-	166.0 s	89.9 s	151.8 d	140.3 s	-	137.8	114.5
					(2.6) d				

a 2_J (CCH). b 1_J. c 3_J (CCCH). d 3_J(CNCH).

1-(Benzylamino)-4-(tert-butylthiocarbamoyle)-4-(dimethylamino)-1-(methylthio) butadiene-2,3-dicarbonitrile (14c, one isomer): Et₂O/CH₂Cl₂ (3:1) as eluent; mp 162°C (Et₂O/CH₂Cl₂); ¹H NMR δ 1.61 (s, 9H), 2.32 (s, 3H), 3.16 (s, 6H), 4.66 (d, J = 6 Hz, 2H), 5.87 (t, J = 6 Hz, 1H), 7.35 (s, 5H), 8.43 (br, 1H); ¹³C NMR δ 17.3 (q, ¹J = 142 Hz), 26.9 (qm, ¹J = 127 Hz), 42.0 (qq, ¹J = 139 Hz, ³J = 3.4 Hz), 49.2 (tm, ¹J = 139 Hz), 57.3 (m), 63.5 (s), 78.6 (d, ³J = 1.5 Hz), 121.0, 122.6 (2s), 127.2 (dd, ¹J = 161 Hz, ³J = 5.5 Hz), 127.5 (dt, ¹J = 161 Hz, ³J = 4 Hz), 128.7 (dm, ¹J = 158 Hz), 138.2, 161.5, 164.0 (3m), 189.9 (s) (Selective irradiation on the dimethylamino group at δ 3.18 causes the C-4 signal at δ 164.0 to turn into a doublet, or into a singlet in the presence of D₂O, and reveals the coupling constant ³J_(CCNH) to be 2.4 Hz); MS calcd for C₂₁H₂₇N₅S₂ m/z 413.1708 (M[†]), found 413.1728; m/z (rel int) 413 (3), 91 (100); IR 3340, 3275, 2160 cm⁻¹. Anal. Calcd for C₂₁H₂₇N₅S₂: C, 61.02; H, 6.54; N, 16.95; S, 15.50. Found: C, 61.04; H, 6.59; N, 16.80; S, 15.31.

1-Benzyl-5-(*tert*-butylthiocarbamoyle)-2-(methylthio)-1H-pyrrole-3,4 dicarbonitrile (15c): Et₂O/petroleum ether (2:1) as eluent; mp 181°C (MeOH); 1 H NMR δ 1.37 (s, 9H), 2.40 (s, 3H), 5.72 (s, 2H), 7.15 (m, 5H), 7.66 (br, 1H); MS calcd for $C_{19}H_{20}N_4S_2$ m/z 368.1129 (M[†]), found 368.1141; m/z (rel int) 368 (23), 91 (100); IR 3278, 2220 cm⁻¹. Anal. Calcd for $C_{19}H_{20}N_4S_2$: C, 61.96; H, 5.43; N, 15.22; S, 17.39. Found: C, 62.18; H, 5.49; N, 15.07; S, 17.48.

1-Benzyl-5-(*tert*-butylthiocarbamoyle)-2-(phenylthio)-1H-pyrrole-3,4 dicarbonitrile (15f): Et₂O/petroleum ether (2:1) as eluent; mp 149°C (MeOH); 1 H NMR 3 1.35 (s, 9H), 5.66 (s, 2H), 7.05 (m, 10H), 7.57 (br, 1H); MS calcd for $C_{24}H_{22}N_4S_2$ m/z 430.1286 (M 4), found 430.1298; m/z (rel int) 430 (27), 167 (72), 91 (100); IR 3290, 2223 cm $^{-1}$. Anal. Calcd for $C_{24}H_{22}N_4S_2$: C, 66.98; H, 5.12; N, 13.02; S, 14.88. Found: C, 66.73; H, 5.46; N, 12.79; S, 14.78.

 $\begin{array}{l} \textbf{5-(tert\text{-}Butylthiocarbamoyle)-1-[(ethoxycarbonyl)methyl]-2-(methylthio)-1H-pyrrole-3,4-dicarbonitrile (15g): Et_2O as eluent; mp 102°C (MeOH); <math display="inline">^1H$ NMR δ 1.27 (t, J=7 Hz, 3H), 1.59 (s, 9H), 2.43 (s, 3H), 4.23 (q, J=7 Hz, 2H), 5.30 (s, 2H), 7.89 (br, 1H); MS calcd for $C_{16}H_{20}N_4O_2S_2$ m/z 364.1028 (M $^+$), found 364.1035; m/z (rel int) 364 (100), 308 (30), 279 (19), 276 (21), 261 (30), 247 (18), 236 (30), 235 (39), 201 (34); IR 3280, 2223, 1750 cm $^{-1}$. Anal. Calcd for $C_{16}H_{20}N_4O_2S_2$: C, 52.75; H, 5.49; N, 15.38; S, 17.58. Found: C, 52.65; H, 5.67; N, 15.33; S, 17.61.

5-tert-Butyl-4-imino-1-methyl-6a-phenyl-2-(phenylthio)-6-thioxo-3a,4,5,6a-tetra-hydro-1H, 6H-pyrrolo [3,4-b] pyrrole-3-carbonitrile (16a): Et₂O/petroleum ether (3:2) as eluent; mp 175°C (MeOH); 1 H NMR δ 1.86 (s, 9H), 2.83 (s, 3H), 3.70 (s, 1H), 7.30 (m, 10H); MS calcd for C₂₄H₂₄N₄S₂ m/z 432.1442 (M⁺), found 432.1433; m/z (rel int) 432 (48), 376 (38), 318 (15), 315 (18), 291 (100), 290 (77), 267 (34); IR 3260, 2187, 1670 cm⁻¹. Anal. Calcd for C₂₄H₂₄N₄S₂: C, 66.67; H, 5.56; N, 12.96; S, 14.81. Found: C, 66.50; H, 5.53; N, 12.80; S, 14.58.

endo-6-(tert-Butylimino)-2,7-diphenyl-2,3,3a,4,7,7a-hexahydro-4,7-epi-(N-methylimino)-4-(phenylthio)-1H, 6H-pyrrolo [3,4-c] thiopyran-1,3-dione (17a): Et₂O/petroleum ether (1:1) as eluent; mp 185°C (MeOH); 1 H NMR δ 1.15 (s, 9H), 2.52 (s, 3H), 3.28, 3.39

(AB syst, J = 9.7 Hz, 2H), 7.65 (m, 15H); MS calcd for $C_{30}H_{29}N_3O_2S_2$ m/z 527.1701 (M⁺), found 527.1709; calcd for $C_{25}H_{20}N_2O_2S$ m/z 412.1245 (M-t-BuNCS)⁺, found 412.1263; m/z (rel int) 527 (0.1), 412 (56), 335 (16), 302 (25), 188 (28), 184 (100); IR 1770, 1712, 1642 cm⁻¹. Anal. Calcd for $C_{30}H_{29}N_3O_2S_2$: C, 68.31; H, 5.50; N, 7.97. Found: C, 68.46; H, 5.46; N, 7.90.

3-[(Benzylamino)-(methylthio)methylene]-4-[(tert-butylthiocarbamoyle)-(dimethylamino)methylene]-1-phenylsuccinimide (19c): Ethyl acetate as eluent; mp 190°C (MeOH); 1 H NMR δ 1.40 (s, 9H), 2.13 (s, 3H), 2.77 (s, 3H), 3.01 (s, 3H), 4.73 (d, J=6 Hz, 2H), 6.88 (br, 1H), 7.30 (m, 10H), 9.31 (t, br, 1H); 13 C NMR δ 17.1 (q, $^{1}J=142$ Hz), 29.7 (qm, $^{1}J=126$ Hz), 38.1 (qq, $^{1}J=138$ Hz), 42.6 (qq, $^{1}J=138$ Hz, $^{3}J=3.3$ Hz), 49.2 (tm, $^{1}J=138$ Hz), 57.8 (m), 98.0, 105.5 (2s), 126.6, 127.1 (2 dm, $^{1}J=160$ Hz), 127.3, 127.4 (2 dt, $^{1}J=159$ Hz, $^{3}J=7.1$ Hz), 128.6, 128.7 (2 dd, $^{1}J=161$ Hz, $^{3}J=7.1$ Hz), 132.7 (t, $^{3}J=7.2$ Hz), 137.4, 138.3, 158.6 (3m), 167.3, 168.0, 187.2 (3s); MS calcd for $C_{22}H_{21}N_3O_2S$ m/z 391.1354 [M-BuNHC(S)H] ‡ , found 391.1367; m/z (rel int) 391 (27), 384 (29), 327 (38), 300 (100); IR 3202, 1692, 1644 cm $^{-1}$. Anal. Calcd for $C_{27}H_{32}N_4O_2S_2$: C, 63.78; H, 6.30; N, 11.02. Found : C, 63.39; H, 6.48; N, 10.75.

4-(Benzylamino)-6-(*tert*-butylimino)-7-(dimethylamino)-2-phenyl-1H, 6H-pyrrolo [3,4-c] thiopyran-1,3 (2H)-dione (20c): Ethyl acetate/pentane (1:1) as eluent; mp 200°C (EtOH); 1 H NMR δ 1.40 (s, 9H), 2.96 (s, 6H), 4.46 (br, 2H), 7.30 (m, 10H), 8.56 (br, 1H); MS calcd for $C_{26}H_{28}N_4O_2S$ m/z 460.1933 (M $^+$), found 460.1919; m/z (rel int) 460 (14), 403 (69), 313 (21), 91 (100); IR 3286, 1720, 1657 cm $^{-1}$. Anal. Calcd for $C_{26}H_{28}N_4O_2S$: C, 67.83; H, 6.09; N, 12.17; S, 6.96. Found: C, 67.99; H, 6.14; N, 12.10; S, 6.66.

6-(tert-Butylimino)-7-(dimethylamino)-2-phenyl-4-(isopropylamino)-1H, 6H-pyrrolo [3,4-c] thiopyran-1,3 (2H)-dione (20j): Et₂O/petroleum ether (2:1) as eluent; mp 136°C (MeOH); 1 H NMR δ 1.27 (d, J = 6 Hz, 6H), 1.46 (s, 9H), 2.95 (s, 6H), 3.77 (m, 1H), 7.40 (s, 5H), 8.13 (br, 1H); MS calcd for $C_{22}H_{28}N_4O_2S$ m/z 412.1933 (M $^{+}$), found 412.1933; m/z (rel int) 412 (21), 355 (100), 329 (18), 311 (23), 296 (15); IR 3250, 1720, 1660 cm $^{-1}$. Anal. Calcd for $C_{22}H_{28}N_4O_2S$: C, 64.08; H, 6.80; N, 13.59; S, 7.77. Found: C, 64.17; H, 6.77; N, 13.49; S, 7.97.

References and Notes

- Shiba, T.; Kato, H. Bull. Chem. Soc. Jpn. 1971, 44, 1864-1868; 1973, 46, 964-967. Grashey, R.; Jaenchen, E.; Litzke, J. Chem. Ztg. 1973, 97, 657-658.
- 2 Berrée, F.; Malvaut, Y.; Marchand, E.; Morel, G. J. Org. Chem. 1993, 58, 6022-6029.
- Exhaustive reviews: Newton, C.G.; Ramsden, C.A. Tetrahedron 1982, 38, 2965-3011. Potts, K.T. Mesoionic Ring Systems. In 1,3-Dipolar Cycloaddition Chemistry; Padwa, A., Ed.; Wiley, J.: New York, 1984, vol. 2, chap. 8, pp. 1-82. Ollis, W.D.; Stanforth, S.P.; Ramsden, C.A. Tetrahedron 1985, 41, 2239-2329.
- 4 Pioneering papers: Bayer, H.O.; Gotthardt, H.; Huisgen, R. Chem. Ber. 1970, 103, 2356-2367. Huisgen, R.; Gotthardt, H.; Bayer, H.O.; Schaefer, F.C. Chem. Ber. 1970, 103, 2611-2624. For recent review, see: Bean, G.P. The Synthesis of 1H-Pyrroles. In Pyrroles; Jones, R.A., Ed.; Wiley, J.: New York, 1990, vol. 48, part one, pp. 165-168.
- 5 Padwa, A.; Lim, R.; Mac Donald, J.G.; Gingrich, H.L.; Kellar, S.M. J. Org. Chem. 1985, 50, 3816-3823.
- 6 Review: Osterhout, M.H.; Nadler, W.R.; Padwa, A. Synthesis 1994, 123-141.
- 7 Barrett, G.C. Tetrahedron 1980, 36, 2023-2058.

- The potentiality of 5-aminothiazolium salts 1 to serve as synthetic equivalents of cyclic azomethine ylides 2 in basic media has only been mentioned in our earlier report 2 where we described the easy preparation of one pyrrole (4b) through the use of DMAD.
- To our knowledge and except the above example, ⁸ extrusion of isothiocyanate from mesoionic compounds was only related for the rearrangement of the 4-methyl-5-phenyl-1,3,4-thiadiazolium-2-anilide into the corresponding 1,3,4-triazolium-2-thiolate. This isomerisation proceeds via a cycloaddition with phenyl isothiocyanate: Ollis, W.D.; Ramsden, C.A. *J. Chem. Soc., Perkin Trans. I* 1974, 633-638.
- 10 Padwa, A.; Burgess, E.M.; Gingrich, H.L.; Roush, D.M. J. Org. Chem. 1982, 47, 786-791. Dalla Croce, P.; La Rosa, C. Heterocycles 1988, 27, 2825-2832 and references cited therein for the works by R. Huisgen, H. Gotthardt et al..
- 11 Texier, F.; Mazari, M.; Yebdri, O.; Tonnard, F.; Carrié, R. Bull. Soc. Chim. Fr. 1991, 128, 962-967
- 12 Similarly, dimethyl maleate rapidly gives isomeric fumarate in the presence of DBN but is quite stable in the presence of NEt3.
- 13 Sustmann, R.; Huisgen, R.; Huber, H. Chem. Ber. 1967, 100, 1802-1813.
- For instance, the mesoionic tetraphenylimidazolium-4-olate was found to be sluggishly reactive towards dimethyl fumarate and maleate giving succinimidopyrrolines via a process like (c), Scheme 5: Singh, G.; Pande, P.S. Tetrahedron Lett. 1974, 2169-2170.
- 15 Friedrichsen, W.; Schröer, W.D. Tetrahedron Lett. 1977, 1603-1604.
- 16 Texier, F.; Yebdri, O.; Laidoudi, A.; Talbi, B.; Balegroune, F.; Germain, G. Tetrahedron Lett. 1983, 24, 189-192.
- 17 Schmitt, G.; Laude, B.; Vebrel, J.; Rodier, N.; Theobald, F. Bull. Soc. Chim. Belg. 1989, 98, 113-123.
- 18 Benages, I.A.; Albonico, S.M. J. Org. Chem. 1978, 43, 4273-4276. Texier, F.; Mazari, M.; Yebdri, O.; Tonnard, F.; Carrié, R. Tetrahedron 1990, 46, 3515-3526 and references therein.
- 19 Gotthardt, H.; Christl, B. Chem. Ber. 1978, 111, 3029-3036.
- 20 Padwa, A.; Hertzog, D.L. Tetrahedron 1993, 49, 2589-2600.
- Similar observations have been made for the vicinal *cis*-coupling constants of 2,3-disubstituted-*endo*-bornane derivatives (${}^3J = 8.8 9.9$ Hz for exo -H-2 and H-3) 22 which are uniformly larger than those of exo-compounds (${}^3J = 7-8.1$ Hz for endo-H-2 and H-3), 23 probably for steric reasons. 24 Likewise, the H-2/H-3 vicinal cis-coupling constants observed in 7-azanorbornanes and 7-azanorbornene complexes are smaller in exo-isomers (endo-ring protons, ${}^3J > 9.6$ Hz) than in endo-isomers (exo-ring protons, ${}^3J > 9.6$ Hz). 25
- Banks, M.R.; Blake, A.J.; Cadogan, J.I.G.; Dawson, I.M.; Gosney, I.; Grant, K.J.; Gaur, S.; Hodgson, P.K.G.; Knight, K.S.; Smith, G.W.; Stevenson, D.E. Tetrahedron 1992, 48, 7979-8006.
- 23 Bonner, M.P.; Thornton, E.R. J. Am. Chem. Soc. 1991, 113, 1299-1308.
- 24 Daniel, A.; Pavia, A.A. Bull. Soc. Chim. Fr. 1971, 1060-1074.
- 25 Gonzalez, J.; Koontz, J.I.; Hodges, L.M.; Nilsson, K.R.; Neely, L.K.; Myers, W.H.; Sabat, M.; Harman, W.D. J. Am. Chem. Soc. 1995, 117, 3405-3421.
- 26 Potts, K.T.; Baum, J.; Houghton, E.; Roy, D.N.; Singh, U.P. J. Org. Chem. 1974, 39, 3619-3627.
- 27 Barrett, G.C.; Walker, R. Tetrahedron 1976, 32, 571-577 and 579-581.